

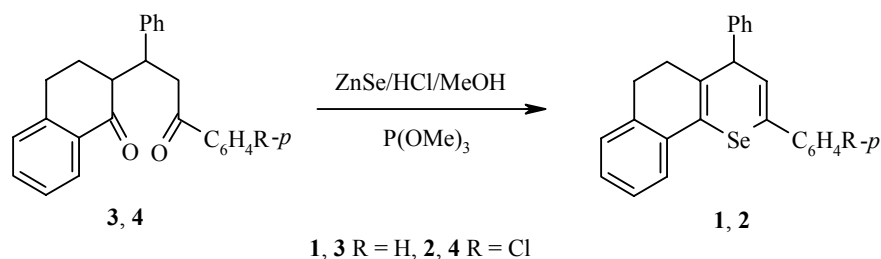
## SYNTHESIS OF THE FIRST REPRESENTATIVES OF BENZANNELATED DIHYDROSELENOCHROMENES

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**Keywords:** 1,5-diketone, zinc selenide, 5,6-dihydrobenzo[*h*]selenochromene.

Benzannelated 5,6-dihydrothiochromenes became available relatively recently [1], but until now their selenium-containing analogs have been unknown.

We have for the first time realized the synthesis of 2,4-diaryl-5,6-dihydrobenzo[*h*]selenochromenes **1** and **2** by the reaction of 2-(1,3-diaryl-3-oxopropyl)tetrahydronaphthalen-1-ones **3** and **4** with hydrogen selenide *in situ* under the conditions of acid catalysis.



The reaction was conducted in the presence of trimethoxyphosphine, produced by the reaction of phosphorus(III) chloride with methanol. Here the necessary concentration of hydrogen chloride (7-8 M) was created, and the secondary disproportionation processes of the dihydrobenzoselenochromenes were suppressed [2]. The yields of the dihydrobenzoselenochromenes **1** and **2** amounted to 91 and 67% respectively.

Chromato-mass spectrometric investigations showed that thermal isomerization of the dihydrobenzoselenochromene **1** occurs under the conditions of GLC (EI 70 eV, 280°C) with the formation of four isomers (molecular ion  $m/z$  400 for the  $\text{Se}^{80}$  isotope). In addition, the chromatogram contains signals for two products with molecular ions  $m/z$  398 and 402 (for  $\text{Se}^{80}$ ), which by analogy with 2,4-diphenyl-5,6-dihydrobenzo[*h*]thiochromene [1] probably result from the disproportionation of compound **1** with the participation of the heterocyclic and alicyclic fragments.

The  $^1\text{H}$  NMR spectra were recorded in  $\text{CDCl}_3$  on a Varian FT-80a instrument (80 MHz) with TMS as internal standard.

**2,4-Diphenyl-5,6-dihydrobenzo[*h*]selenochromene (1).** To methanol (20 ml), placed in a flask with the minimum amount of free space, we added with constant stirring and cooling  $\text{PCl}_3$  (9.42 g, 68 mmol), compound **3** (1.42 g, 4 mmol), and ZnSe (0.72 g, 5 mmol). The mixture was stirred at room temperature for 7 h. The separated crystals were removed, washed with ethanol, and dried. We obtained 1.27 g of compound **1**. The

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mixture was extracted with diethyl ether (50 ml), the extract was washed with water to a neutral reaction and dried, and the solvent was evaporated. The oily product was passed through a column of Al<sub>2</sub>O<sub>3</sub> (eluent 10:1 hexane–ether). A further 0.17 g of compound **1** was obtained (total yield 91%); mp 108-109°C (reprecipitated from diethyl ether with ethanol). <sup>1</sup>H NMR spectrum, δ, ppm (*J*, Hz): 6.87-7.49 (14H, m, H<sub>Ar</sub>); 6.30 (1H, d, *J* = 5.8, β-H); 4.37 (1H, d, *J* = 5.8, γ-H); 2.88-2.67 (2H, m, CH<sub>2</sub>); 2.38-2.19 (2H, m, CH<sub>2</sub>). Found %: C 74.95; H 5.27. C<sub>25</sub>H<sub>20</sub>Se. Calculated %: C 75.19; H 5.01.

**2-(4-Chlorophenyl)-4-phenyl-5,6-dihydrobenzo[*h*]selenochromene (2)**. This compound was obtained similarly to compound **1**, and the reaction took 37 h. Yield 67%; mp 126-128°C (reprecipitated from diethyl ether with ethanol). <sup>1</sup>H NMR spectrum, δ, ppm (*J*, Hz): 7.05-7.49 (13H, m, H<sub>Ar</sub>); 6.24 (1H, d, *J* = 5.4, β-H); 4.34 (1H, d, *J* = 5.9, γ-H), 2.86-2.49 (2H, m, CH<sub>2</sub>); 2.35-1.97 (2H, m, CH<sub>2</sub>). Found %: C 69.03; H 4.66. C<sub>25</sub>H<sub>19</sub>ClSe. Calculated %: C 69.20; H 4.38.

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